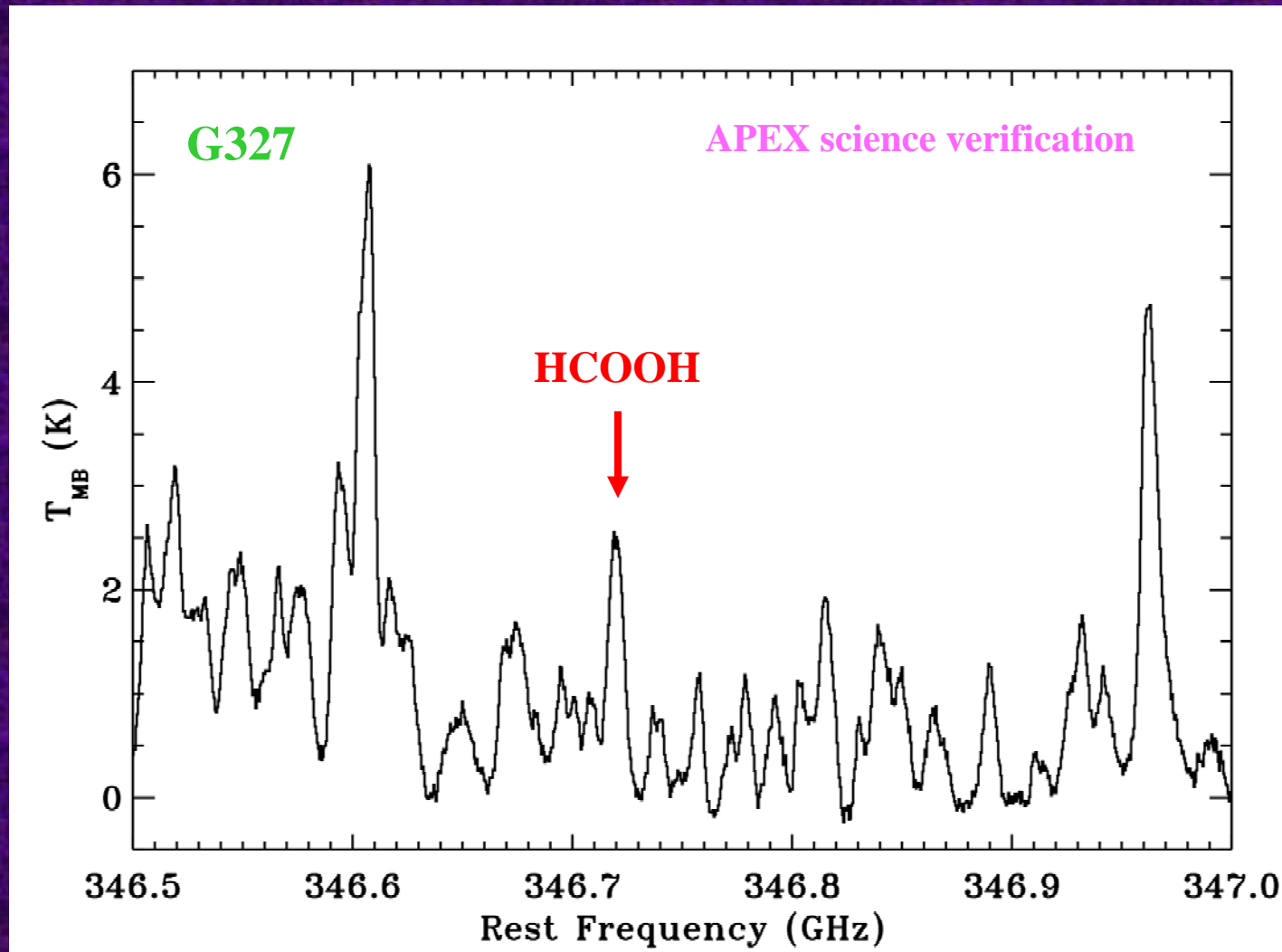


Testing grain-surface chemistry in massive hot-core regions

Suzanne Bisschop
Jes Jørgensen
Ewine van Dishoeck
09-05-06

Introduction I

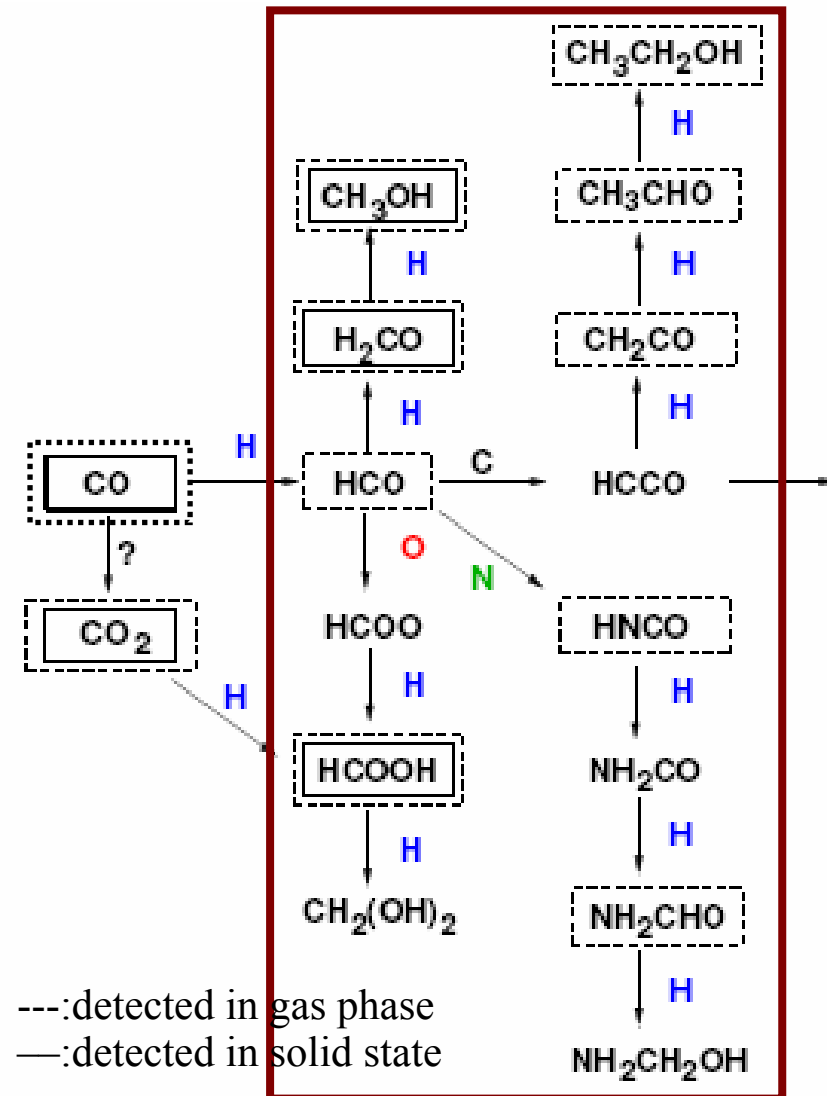


High mass star forming regions rich in organics!

Introduction II

- 2 proposed formation mechanisms:
 - Grain-surface chemistry
 - High temperature gas phase chemistry

Based on Tielens and Charnley 1997



Observations JCMT & IRAM

- 7 high mass YSOs
- Source selection criteria:
 - Lines $< 5 \text{ km s}^{-1}$
 - $D < 5 \text{ kpc}$
 - $T_a > 1 \text{ K}$ 338 GHz $7_K - 6_K$ CH₃OH branch
- Luminosities $10^4 - 10^6 L_\odot$
- Solid state data available

AFGL 2591

G24.78

G75.78

N6334 IRS1

N7538 IRS1

W 3(H₂O)

W 33A



- Line selection criteria:
 - Low + high lying energy levels
 - Not close to transitions other molecules
 - Isotopes H₂CO, CH₃OH, and HNCO
 - IRAM: low lying energy levels of CH₂CO, CH₃CHO, and HCOOH



Goal 1

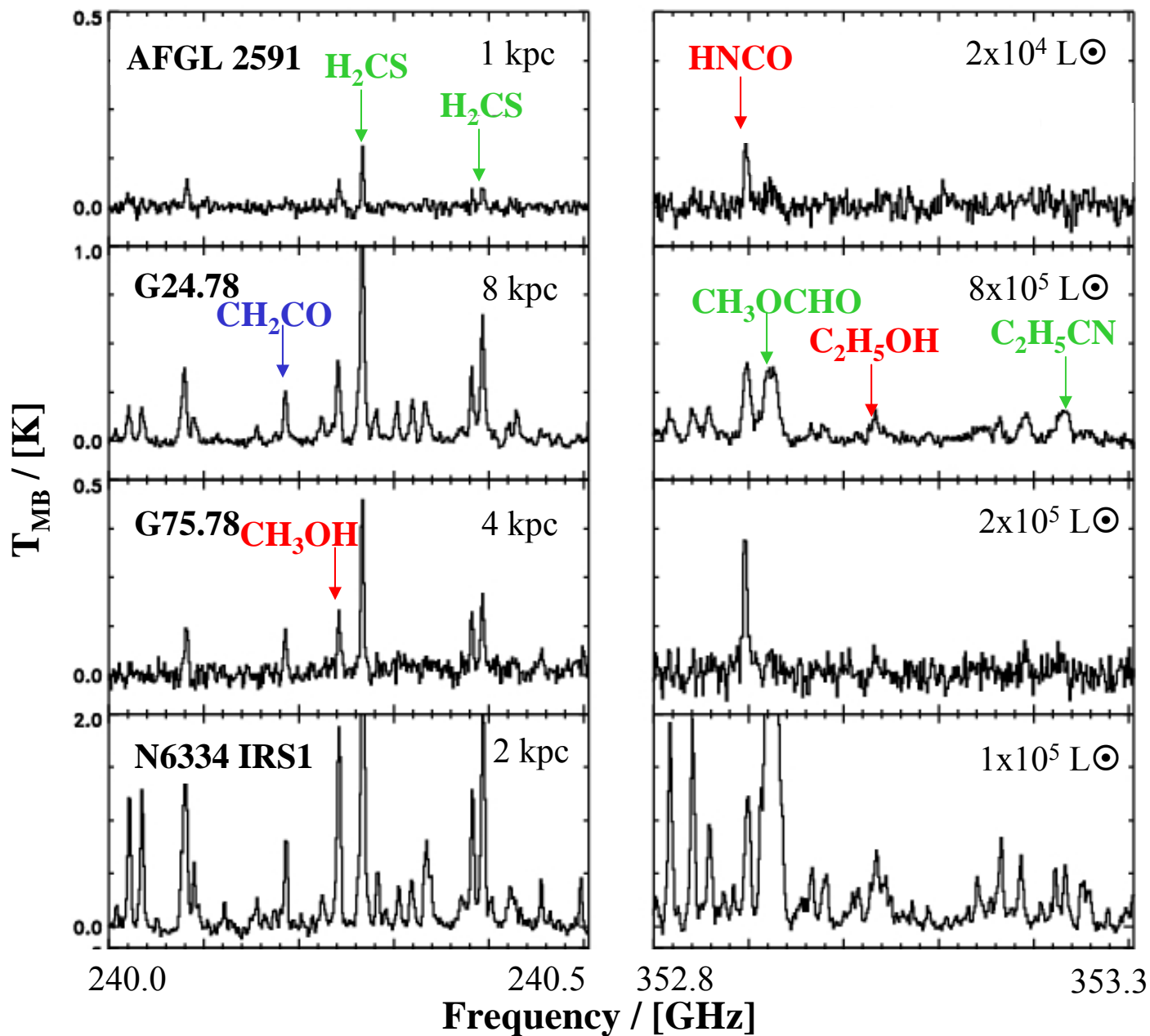
- Rotation temperatures: hot or cold?
- Inventory abundances grain-surface species



Link molecules chemically

Results: spectra

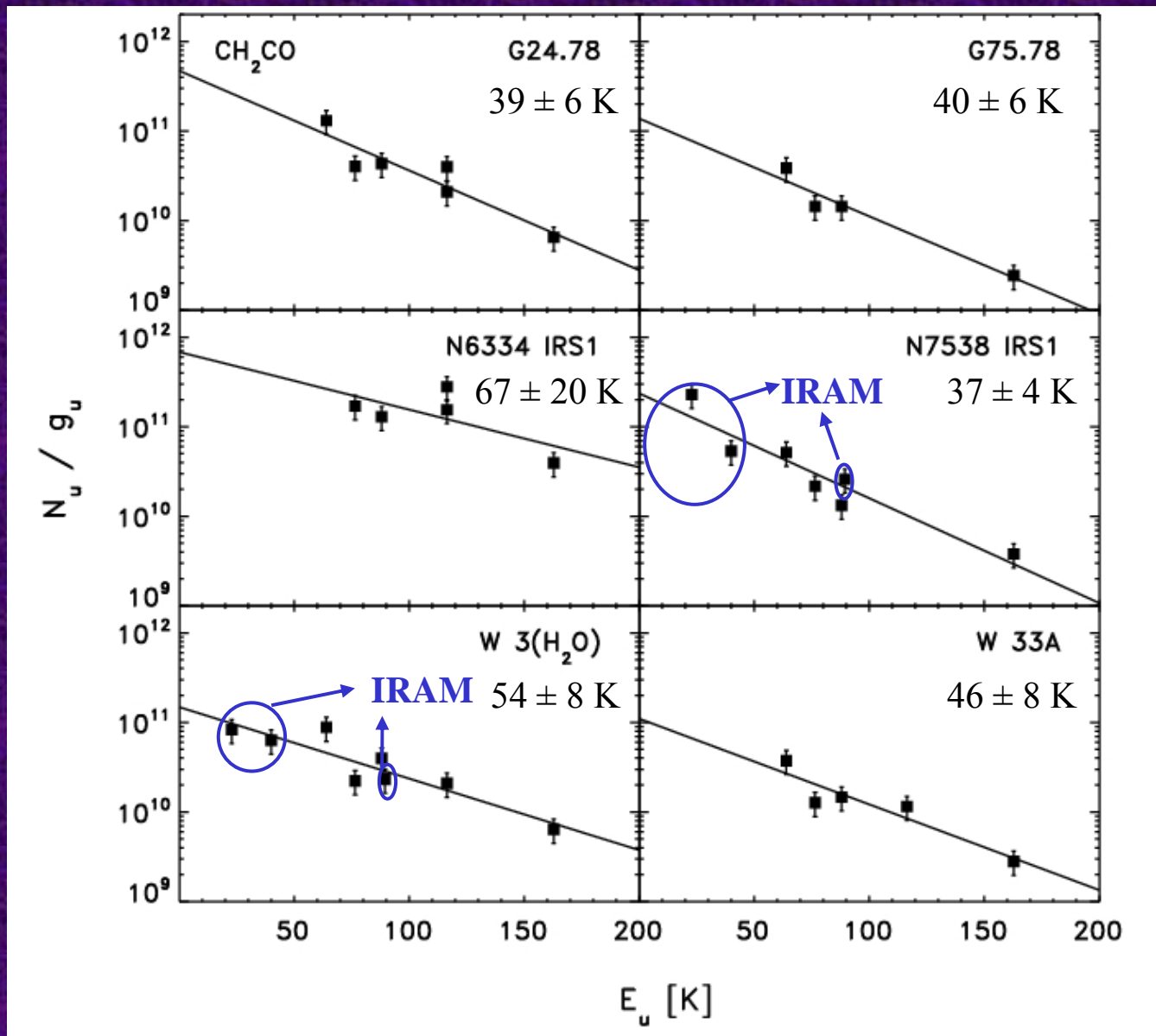
Varying line richness !



Analysis

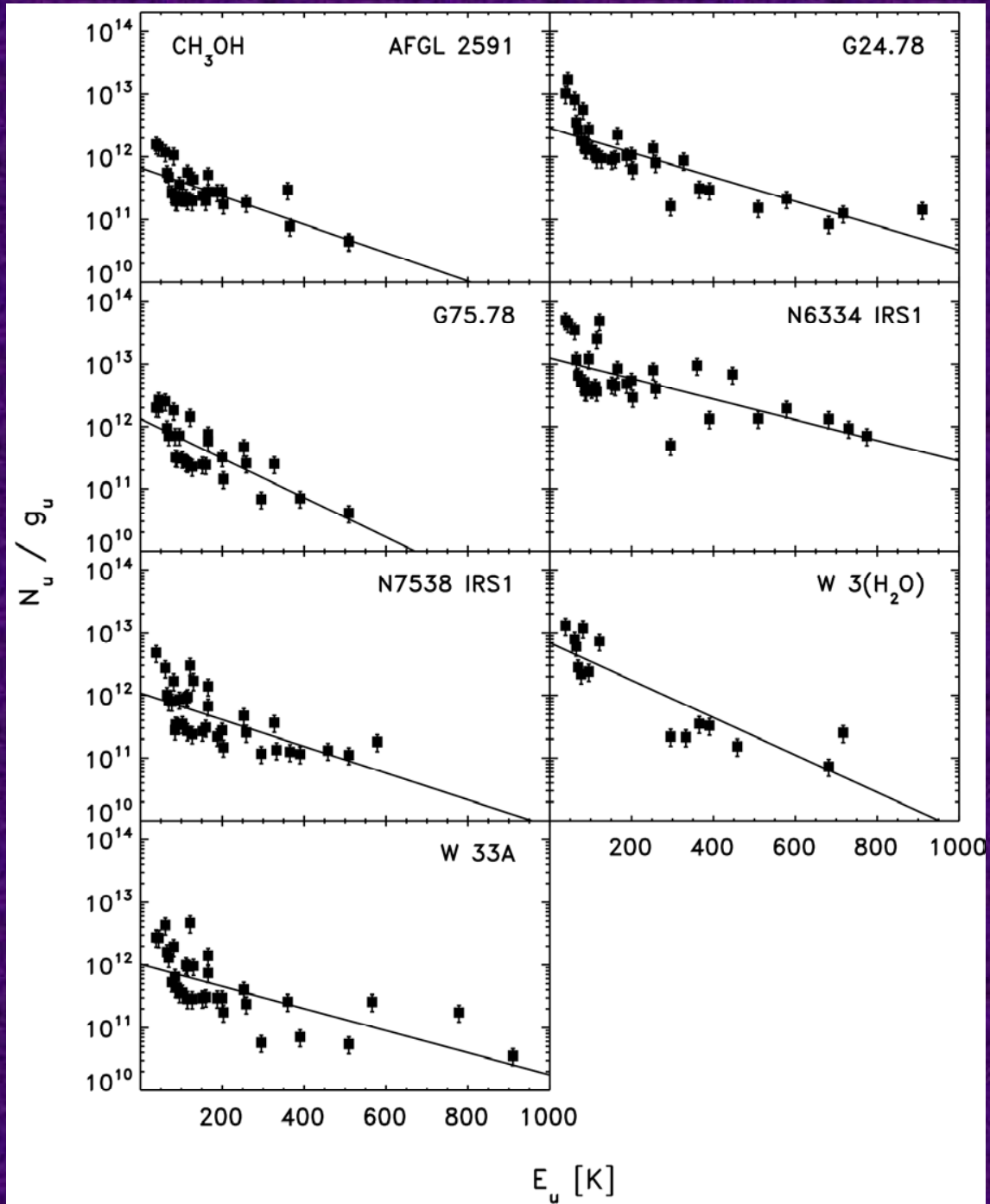
- Rotation diagram
 - T_{rot} & N
- Dust radiative transfer models based on 850 μm SCUBA data + L
 - $R_{T=100\text{ K}} \rightarrow$ beam-dilution: $N(X)/bf$
 - $N(\text{H}_2)_{T>100\text{ K}} \rightarrow$ abundances: $N_{bf}(X)/N(\text{H}_2)$
- Correlation abundances

Rotation diagrams CH_2CO

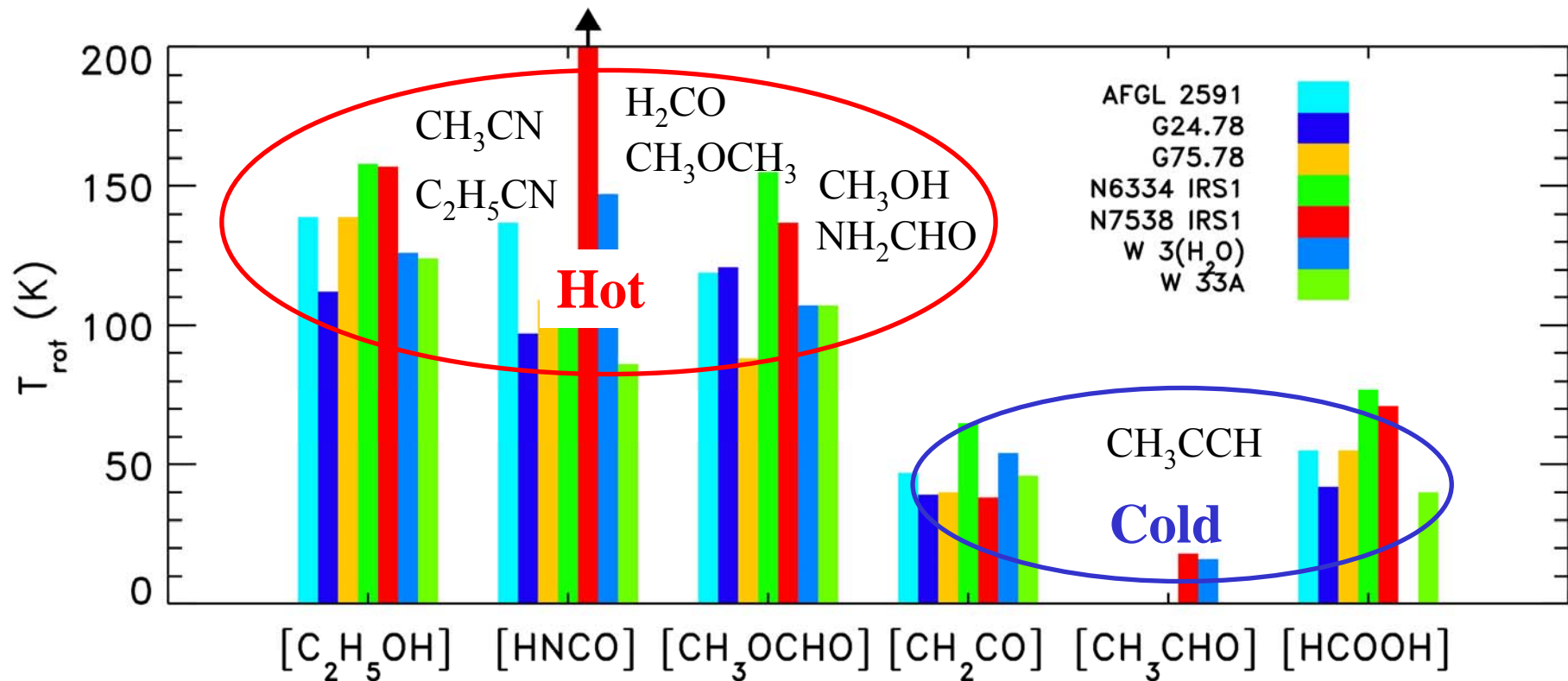


Assumptions:
emission coming
from a region
with a single
temperature and
emission
optically thin

Rotation diagrams: CH₃OH



Rotational temperatures



Beam-dilution and $N(\text{H}_2)_{T>}$

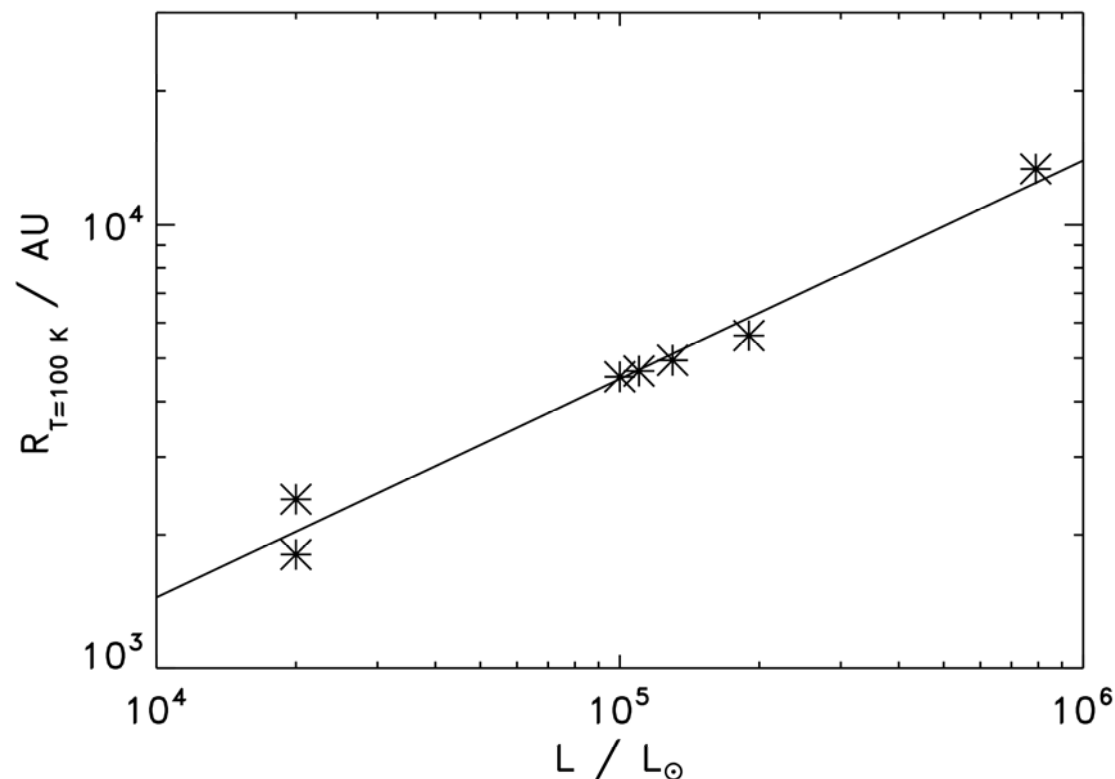
100 K

- $R_{T=100\text{ K}}: 10^3\text{-}10^4$ AU
- $N_{T>100\text{ K}}: 10^{22}\text{-}10^{23}$ cm⁻²
- $Bf: 0.02\text{-}0.1$

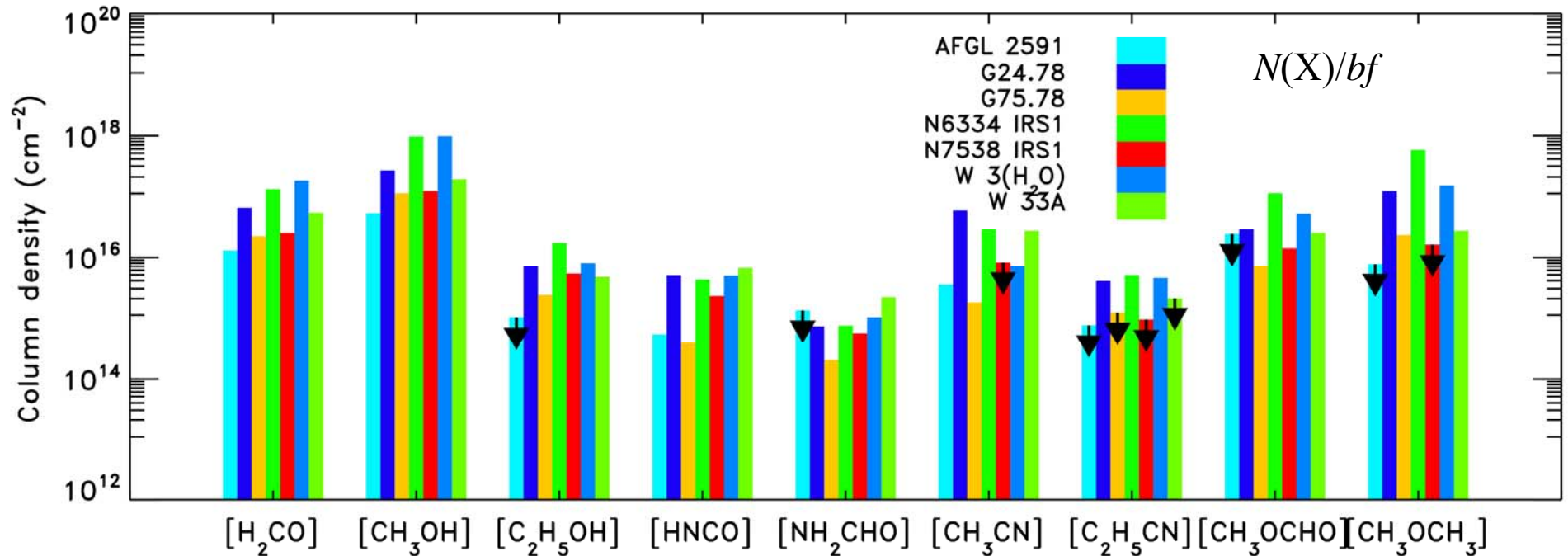
Aim:

• $N(X)/bf$: source averaged column densities

• $N_{bf}(X)/N(\text{H}_2)_{T>100\text{ K}}$: abundances

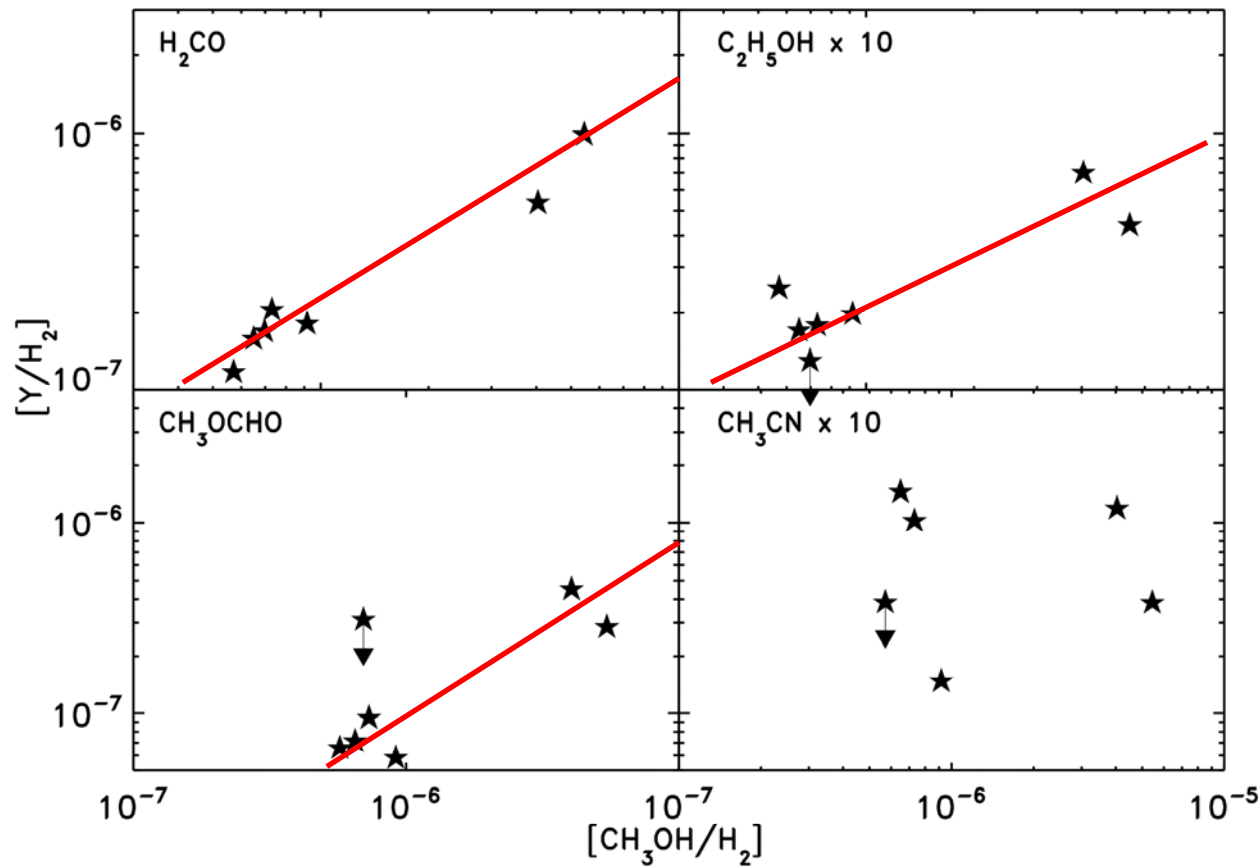


Column density trends (hot gas)



- Similar column density variations (1-2 orders of mag.) for
 - H₂CO, CH₃OH, C₂H₅OH, CH₃OCH₃, and CH₃OCHO
 - HNCO, NH₂CHO
- Note independent of beam-dilution correction!

Correlation abundances with respect to CH_3OH

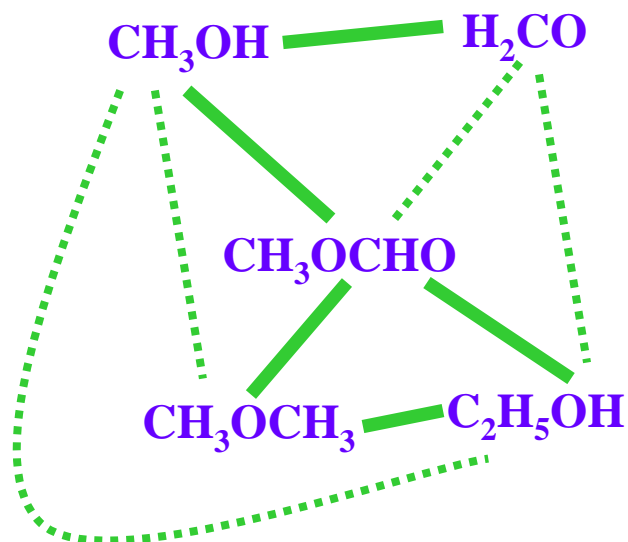


- Abundances of molecules compared in 100 K gas
 - O-bearing species correlate
 - N-bearing species correlate (HNCO & NH_2CHO)

Branching ratios

	Ratio	σ
H ₂ CO/CH ₃ OH	0.21	0.05
C ₂ H ₅ OH/CH ₃ OH	0.025	0.013
CH ₃ OCHO/CH ₃ OH	0.098	0.032
CH ₃ OCH ₃ /CH ₃ OH	0.31	0.20
NH ₂ CHO/HNCO	0.27	0.14

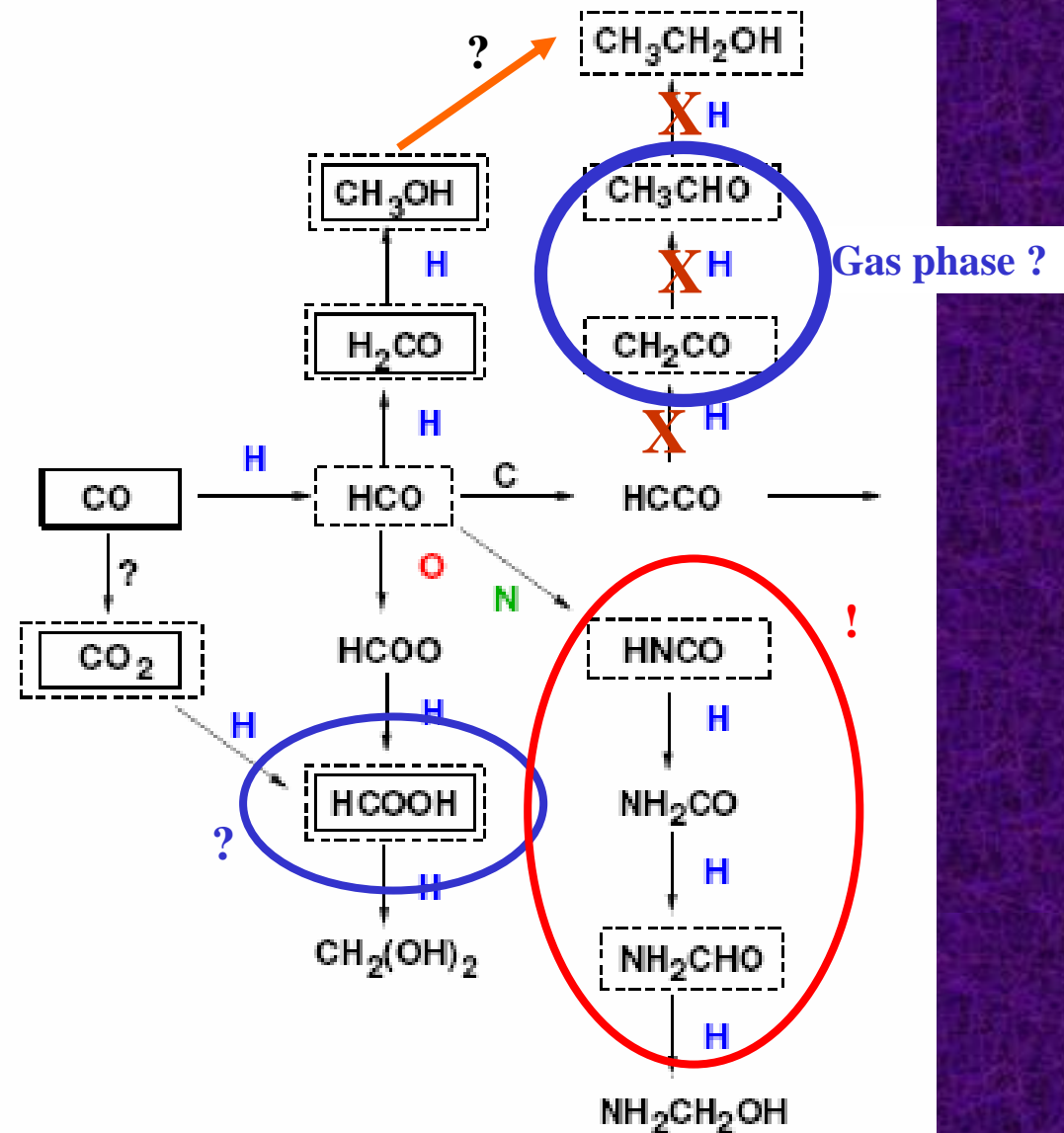
- Correlating species have constant branching ratios



Implications

- Hydrogenation to C_2H_5OH very efficient?/ Route through CH_3OH ?
- Formation CH_2CO and CH_3OH ?
- $HCOOH$ relation unclear
- $HNCO$ and NH_2CHO related

Based on Tielens and Charnley 1997



Conclusion

S

- Some molecules always hot, others always cold
- Significant column density (abundance) variations in hot gas
- Strong correlations for O-bearing and N-bearing species, but not with each other!

Future work:

- Spatially resolved observations
- Laboratory experiments