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LondonHow do conical intersections control
photostability and photochemical reactivity?

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Contents

1. Important concepts in photochemistry: role of conical intersections and their topology in photostability and photoreactivity

- Notion of peaked / sloped conical intersection
- Importance of reaction path in non-adiabatic processes
- 2. Illustrations:
 - DHA/VHF photochromism
 - Photoactivation of the Photoactive Yellow Protein (PYP)



Degradation through a peaked intersection

- Peaked intersection: crossing point is lowest energy point on excited
- Several channels arising on ground state from the crossing.
- Even if quantum yields of photoproducts A and B are small, over many photocycles the reactant will degrade.
- Conclusion: accessible peaked intersection is bad for photostability.

Degradation Channel B

→ Photoproduct

Degradation Channel A

 \rightarrow Photoproduct

Photostability through a sloped intersection



• Sloped intersection: lower energy excited state minimum than crossing point.

• Only one relaxation channel on ground state leading back to reactant.

• Vibrationally "hot" reactant M can be regenerated on an ultrafast timescale if the energy gap between M* and the CI is small.

• Model case for photostability.

3 essentials coordinates to understand non-adiabatic processes:

- non-adiabatic coupling vector: q_1
- gradient difference vector: q_2
- reaction path: Rx





branching space characterising the crossing

<u>2 cases:</u> ① Rx in the branching space:

- Crossing appears as a double-cone.
- System relaxes in the upper cone and decays near the tip of the cone (hourglass-like funnel).
- Expect the system to decay near the lowest energy crossing point.

\bigcirc Rx \perp to the branching space:

- Crossing appears as a seam of intersection.
- Vibrational motion orthogonal to the reaction path and included in the branching space takes the system to the seam where it can decay.

• Expect non-radiative decay to take place over a large section of the seam.

Delicate interplay between the Rx coordinate and the intersection seam



M. Boggio-Pasqua et al., J. Phys. Chem. A 107, 2003, 11139.



- fluorescence from DHA
- no fluorescence from VHF
- DHA -> VHF: photochemical reaction
- VHF -> DHA: thermal backward reaction
- Φ DHA->VHF \approx 1 at room temperature!!!





The S₁ DHA \rightarrow VHF pathway

- Minimum **M*** accounts for fluorescence from DHA.
- Transition state **TS*** accounts for competition between photochemical ring-opening reaction and fluorescence.
- Conical intersection **X** accounts for ultrafast non-radiative decay.

The S₁ VHF \rightarrow VHF pathway

- Crossing **X** is lowest energy point on S_1 surface.
- No excited state minimum on VHF side: accounts for lack of fluorescence from VHF.
- No relaxation pathway leading back to DHA on S_0 after decay at crossing X: accounts for one-way photochromism.

Photoactive Yellow Protein

<u>Groningen: Gerrit Groenhof</u>, Alan E. Mark et al; Imperial/KCL Mathieu Bouxin-Cademartory, Michael Robb

Sienna: Massimo Olivucci

• negative photo-tactic response to blue light:



PYP protein:

Trans-to-cis photoisomerization of the chromophore:







torsion *b* = *trans*-to-*cis* isomerization coordinate, skeletal deformation = stretching of the bonds

- In vacuo, the S₁ barrier to photoisomerization is too high and the system is trapped in a *trans* minimum on S₁ before decaying back by fluorescence.
- In the protein, the S₁ barrier decreases and the system can relax to a global twisted minimum on S₁ before reaching the conical intersection and decaying to S₀.
- Non-radiative decay occurs along the crossing seam depending on vibration orthogonal to torsion *b* (torsion *b* is orthogonal to branching space).
- The electrostatic field of the protein stabilizes the excited state of the twisted chromophore moving the seam closer to the twisted S₁ minimum.
- Proton transfer becomes favorable in the *cis* isomer and induces the unfolding of the protein.

G. Groenhof et al., J. Am. Chem. Soc. 126, 2004, 4228.

Conceptual basis of Mechanistic Photochemistry

- Radiationless decay via conical intersections.
- CI topography (peaked *vs.* sloped) controls photophysical *vs.* photochemical outcome (DHA/VHF).
- Surface topology (eg. barrier separating FC region from CI?) controls competition between radiative and non-radiative processes (diarylethenes).
- Reaction path direction *vs.* funnel coordinates can control product formation (diarylethenes, PYP).